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- There are some popping noises when heating and a smell is produced (water boiling smell).
- When mixed, the mixture is a transparent blue.
- It mixes as the water boils and becomes slightly darker and less transparent.
- After the liquid evaporates, there is a blue deposit left in the test tube. Some of it seemed to turn white. I think it is left over copper sulfate.

Experiment B

- HCl was clear, magnesium was metallic.
- When combined, it was exothermic, let off vapor, and bubbled.

Experiment C

- Water is clear, ammonia is clear (smells bad), Epsom salt is white.
- mixing Epsom salt has no reaction.
- Mixing in ammonia makes a milky, cloudy mixture with things floating inside.

Experiment D

- Silver nitrate and sodium chloride are clear.
- When mixed, it turns white and begins to deposit a white substance. Cloudy.

Experiment E

- Water is white clear and 32°C.
- Sodium bicarbonate is white powder.

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